CARBONISATION OF COAL

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Some of the organic components of Coal have **Fusion Properties** (**REACTIVE**) when heated, while others donot (**Non-Fusible** or **INERT**).

The ability of a Coal to **swell and melt on being heated** and to **form <u>a Coherent Residue</u>** on cooling is called **CAKING.**

Therefore, Coal is called CAKING or NON-CAKING depending on its action when heated in absence of air. When coal softens and solidifies into a Cake, under such heating conditions, it is CAKING or COKING Coal, but when it crumbles or forms a weakly coherent mass it is called NON-CAKING Coal.

<u>CARBONISATION – the destructive heating of coal</u> <u>in the absence of air, with the production of COKE</u> <u>and the evolution of VOLATILE PRODUCTS.</u>

COKING & NON COKING COALS :

Lignites / brown coals and **Long-Flame Bituminous coals** are two classes of Coals that are **NON-COKING** Coals have properties like – Immature Coals, Non-Coherent nature, Vol. Matter - > 40% & F.C.-<60% . The **Anthracites** are also Non-coking coals giving off little or no Vol. Matter when heated upon.

However, **Bituminous Coals** are the **COKING COALS** with properties like –Mature Coals, Coherent, Fused nature (of Coke), Vol. Matter (20 - 40%).

COKE MAKING - COAL CARBONISATION :

- Coking coals are the coals which when heated in the absence of air, first melt, go in the plastic state, swell and resolidify to produce a solid coherent mass called <u>coke</u>.
- When coking coal is heated in absence of air, a series of physical and chemical changes take place with the evolution of gases and vapours, and the solid residue left behind is called <u>coke</u>.
- Conventional coke making is done in a coke oven battery of ovens sandwiched between heating walls.
- They are carbonised at a temperature around 1000°-1100° C upto a certain degree of devolatization to produce metallurgical coke of desired mechanical and thermo-chemical properties.

Objectives of Carbonisation –

 To produce Carbon-rich product that will provide heat for the Blast Furnace – by removal of Volatile Matter i.e.

Devolatilisation

- To produce Carbon Monoxide for the reduction of the Iron Ore.
- ⇒ During carbonisation, coking coals undergo transformation into plastic state at around 350°-400° C swell and then resolidify at around 500°-550° C to give semi-coke and then coke.

Depending upon the temperature range to which raw coal is subjected to during the combustion process, carbonization is of

2 types -

- 1. Low Temperature Carbonisation -400-800 deg.C
- 2. High Temperature Carbonisation 900–1300 deg.C

LOW TEMPERATURE CARBONISATION

(400 – 800 degrees C)

Objective :- Production of Semi-Coke (Primary Objective) ;

especially suitable for **Domestic Fuel or Boiler Use.**

Products :-

- 1. Semi-Coke or Char
- 2. Tar -- the most important By-product of Low Temp. Carbonisation
- 3. Light Oil
- 4. Other products -- some Gases, Benzene and Tar acids

HIGH TEMPERATURE CARBONISATION

- In this process, the **chief aim** is to have **very little Volatiles** in the Coke.
- Generally, **the V.M. content** of Cokes formed by the process ranges from **2.5-3.0 %**.
 - Coke is **coherent, hard, dense, silver grey** in colour and can neither be ignited nor crushed easily.
 - Product is known as "Hard Coke"
 - Production of Metallurgical (Hard) Coke through hightemperature carbonization process can be through :
 - 1. Bee-hive Coke-oven Process (An older process which had a total loss of By-Products, later replaced by)
 - 2. By-product Coke-oven Process

By-product Coke-oven Process

Construction :-

 OVENS – which are long, narrow insulated refractory structures with removable doors at each end and charging ports in the roof

- Ovens are arranged in **batteries** of **30-100 ovens** side by side
- Coking through Fuel gas circulating through Heating flues in walls (1150- 1350 deg. C)
- Capacity of Each Oven 4-20 tons per charge
- Coking Process :- 13-48 hrs



Fig: Schematic Diagram of Coke Oven Battery

COKE-OVEN BY-PRODUCTS

Carbonisation -- produces **Metallurgical Coke** (as Main Product) **Principal By-products :**

- 1. Coal Gas (Mixture high in Hydrogen + Methane; some CO)
- **2. Ammoniacal Liquor** converted by reaction with H2SO4 to Ammonium Sulphate (used in Fertiliser or as Fire retardant)
- Crude Benzol (C6H5OH) mixture of Benzene, Xylene & Toluene; used in Plastic Industry)
- 4. Coal Tar